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where R is the ideal gas constant, n the molar number, Φ the volume fraction, and the subscripts 1 and 2 denote the solvent and the polymer, respectively. With this equation, ΔS_{com} is always positive because $\Phi_1 + \Phi_2 = 1$ and $\partial \Delta S_{com} / \partial n_2$, the partial molar entropy of the polymer, is also positive when the amount of the solvent is much larger than the polymer.

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4 8 16 In the first call to the function, we only define the argument a, which is a mandatory, positional argument.In the second call, we define a and n, in the order they are defined in the function.Finally, in the third call, we define a as a positional argument, and n as a keyword argument.. If all of the arguments are optional, we can even call the function with no arguments.

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ICMT's PhD student Fazlollah Madani Sani won the Gustavus Edward and Gladys V. Smith Memorial Engineering Award (a fellowship of \$7000) as the 2019 outstanding full-time Chemical Engineering PhD student in the Department of Chemical and Biomolecular Engineering at Ohio University.

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