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Kaksoissidos koostuu sigma- ja pi-sidoksesta (σ ja π), ja kummassakin sidoksessa on kaksi elektronia. Delokalisaatio onnistuu vain, jos π -sidokset ovat samassa tasossa. Vain vierekkäisiin, mutta ei samaan hiileen liittyvä π -sidokset on samassa tasossa. Tämä onnistuu esimerkiksi niin, että nimellisesti joka toinen sidokset on kaksoissidos.

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